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#### 2009-2010 PROJECTS

SE ALASKA - EXHUMATION OF THE COAST MOUNTAINS BATHOLITH DURING THE GREENHOUSE TO ICEHOUSE TRANSITION IN SOUTHEAST ALASKA: A MULTIDISCIPLINARY STUDY OF THE PALEOGENE KOOTZNAHOO FM.

Faculty: Cameron Davidson (Carleton College), Karl Wirth (Macalester College), Tim White (Penn State University) Students: Lenny Ancuta, Jordan Epstein, Nathan Evenson, Samantha Falcon, Alexander Gonzalez, Tiffany Henderson, Conor McNally, Julia Nave, Maria Princen

COLORADO – INTERDISCIPLINARY STUDIES IN THE CRITICAL ZONE, BOULDER CREEK CATCHMENT, FRONT RANGE, COLORADO.

Faculty: David Dethier (Williams) Students: Elizabeth Dengler, Evan Riddle, James Trotta

WISCONSIN - THE GEOLOGY AND ECOHYDROLOGY OF SPRINGS IN THE DRIFTLESS AREA OF SOUTHWEST WISCONSIN.

Faculty: Sue Swanson (Beloit) and Maureen Muldoon (UW-Oshkosh) Students: Hannah Doherty, Elizabeth Forbes, Ashley Krutko, Mary Liang, Ethan Mamer, Miles Reed

# OREGON - SOURCE TO SINK – WEATHERING OF VOLCANIC ROCKS AND THEIR INFLUENCE ON SOIL AND WATER CHEMISTRY IN CENTRAL OREGON.

Faculty: Holli Frey (Union) and Kathryn Szramek (Drake U.)

Students: Livia Capaldi, Matthew Harward, Matthew Kissane, Ashley Melendez, Julia Schwarz, Lauren Werckenthien

#### MONGOLIA - PALEOZOIC PALEOENVIRONMENTAL RECONSTRUCTION OF THE GOBI-ALTAI TERRANE, MONGOLIA.

Faculty: Connie Soja (Colgate), Paul Myrow (Colorado College), Jeff Over (SUNY-Geneseo), Chuluun Minjin (Mongolian University of Science and Technology)

Students: Uyanga Bold, Bilguun Dalaibaatar, Timothy Gibson, Badral Khurelbaatar, Madelyn Mette, Sara Oser, Adam Pellegrini, Jennifer Peteya, Munkh-Od Purevtseren, Nadine Reitman, Nicholas Sullivan, Zoe Vulgaropulos

## KENAI - THE GEOMORPHOLOGY AND DATING OF HOLOCENE HIGH-WATER LEVELS ON THE KENAI PENINSULA, ALASKA

Faculty: Greg Wiles (The College of Wooster), Tom Lowell, (U. Cincinnati), Ed Berg (Kenai National Wildlife Refuge, Soldotna AK) Students: Alena Giesche, Jessa Moser, Terry Workman

#### SVALBARD - HOLOCENE AND MODERN CLIMATE CHANGE IN THE HIGH ARCTIC, SVALBARD, NORWAY.

Faculty: Al Werner (Mount Holyoke College), Steve Roof (Hampshire College), Mike Retelle (Bates College) Students: Travis Brown, Chris Coleman, Franklin Dekker, Jacalyn Gorczynski, Alice Nelson, Alexander Nereson, David Vallencourt

#### UNALASKA - LATE CENOZOIC VOLCANISM IN THE ALEUTIAN ARC: EXAMINING THE PRE-HOLOCENE RECORD ON UNALASKA ISLAND, AK.

Faculty: Kirsten Nicolaysen (Whitman College) and Rick Hazlett (Pomona College) Students: Adam Curry, Allison Goldberg, Lauren Idleman, Allan Lerner, Max Siegrist, Clare Tochilin

#### Funding Provided by: Keck Geology Consortium Member Institutions and NSF (NSF-REU: 0648782) and ExxonMobil

### Keck Geology Consortium: Projects 2009-2010 Short Contributions – OREGON

#### WEATHERING OF A VOLCANIC LANDSCAPE: THE GEOCHEMISTRY OF THE DESCHUTES RIVER WATERSHED, CENTRAL OREGON.

Project Faculty: *HOLLI FREY*, Union College & *KATHRYN SZRAMEK*, Drake University

#### CHEMICAL WEATHERING IN THE DESCHUTES BASIN: HOW WATERSHED FEATURES EFFECT CATION CONCENTRATIONS IN WATER CHEMISTRY

*LIVIA CAPALDI*: Oberlin College Research Advisor: Steven Wojtal

### SPATIAL VARIABLILTY OF TEPHRA SOIL ON DIFFERING GEOMORPHIC SURFACES WITHIN THE HEADWATERS OF THE DESCHUTTES RIVER, OREGON

*MATTHEW HARWARD*: University of North Carolina at Charlotte Research Advisor: Dr. Martha C. Eppes

### INCIPIENT WEATHERING IN SILICIC ROCKS INDICATED BY ENRICHMENT OF REE AND TRACE ELEMENT CONCENTRATIONS: THE HIGH CASCADES, OREGON

*MATTHEW KISSANE*: Union College Research Advisor: Holli Frey

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ASHLEY MELENDEZ: California State University, Fullerton Research Advisor: Brandon Browne

#### INFLUENCE OF CLIMATE AND LITHOLOGY ON SPRING CHEMISTRY IN THE UPPER DESCHUTES RIVER, OREGON

*JULIA SCHWARZ:* Carleton College Research Advisor: Cameron Davidson

#### ALKALINITY AND DISSOLVED ORGANIC CARBON IN SURFACE WATERS OF THE DESCHUTES DRAINAGE BASIN, OREGON

*LAUREN WERCKENTHIEN:* DePauw University Research Advisor: Dr. James G. Mills, Jr.

Funding provided by: Keck Geology Consortium Member Institutions and NSF (NSF-REU: 0648782)

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# PLAGIOCLASE WEATHERING WITH DISTANCE FROM VOIDS IN VOLCANIC ROCKS OF THE DESCHUTES BASIN, CENTRAL OREGON

#### ASHLEY MELENDEZ

California State University, Fullerton Research Advisor: Brandon Browne

### INTRODUCTION

The geology of the Deschutes Basin of Central Oregon is dominated by volcanic rocks. The rocks in this project are Quaternary in age ranging from the Holocene to the Pleistocene. The composition ranges from basalt to rhyolite, but all contain plagioclase. Plagioclase is a good index mineral because the composition includes mobile elements (Na, Ca) that leave the mineral fairly readily during chemical weathering as well as immobile elements (Al) that do not. Plagioclase is generally not the first mineral to display evidence of weathering in volcanic rocks. Volcanic glass, olivine and pyroxene will all begin to weather before plagioclase (Eggleton et al., 1987). There have been many previous works done regarding the weathering of plagioclase in volcanic rocks because plagioclase is a very common mineral, but most of this work has been in warm, wet environments (Dorn and Brady, 1995; Velbel and Losiak, 2008).

The field area for this project focuses on the Deschutes Basin in the Oregon High Desert east of the Cascades (Fig. 1). The Deschutes Basin is a semidesert, receiving less than 30 cm of rain annually (www.noaa.gov). The samples collected include rocks from basaltic to rhyolitic in composition taken from flow fronts as well as from domes. The most important constraint to dissolution of rock into a watershed is the amount of precipitation received in the area. This constraint can account for 58-71% of dissolved ions found in river runoff (Bluth and Kump, 1993). In the dry, semi-desert environment of the Deschutes Basin, this constraint does not have as much as an effect on the rocks as it does on the rocks in the wet climate of the Cascades to the West or on rocks in tropical regions. Weathering does, however vary with distance from cracks, vesicles and pore space in the rock (Eggleton et al., 1987). Void spaces allow water to get into the rock and begin to alter the minerals. When reactions occur between the water and the elemental constituents of the minerals, some of these ions are carried out of the rock and into the watershed. This paper discusses the weathering of plagioclase in the Holocene to the Pleistocene rocks of the study.

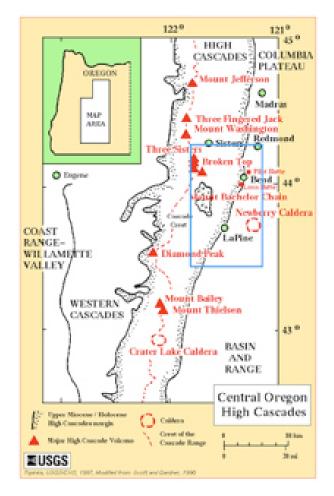


Figure 1: Map of the Deschutes Basin of Central Oregon. Field area is outlined.

## **METHODS**

Forty-eight samples were collected from 22 unique rock outcrops. Rocks were collected from the surface of an outcrop where a higher degree of weathering was expected and from a more central part of the same feature where the rocks were expected to be fresh. Locations of specific rock types were determined using a geological map of the area (Sherrod and Smith, 2000). Thin sections of the samples were prepared and examined by a Nikon Eclipse LV100 POL petrographic microscope in polarized light. Thin sections with cracks or vesicles were of special interest and were examined for evidence of whole rock weathering. The alteration of the susceptible volcanic glass in the samples turns the glass to palagonite and smectie (Glassman, 1982). The altered glass becomes discolored, turning yellow or brownish. The alteration of pyroxenes causes the crystal surfaces to appear etched (Glassman, 1982). While these general indicators of weathering were examined for evidence of general weathering, plagioclase crystals were observed specifically. Plagioclase crystals were checked for evidence of alteration including a decrease in birefringence, yellowish brown discoloration, scalloped rims on the individual phenocrysts, and dusty textures within crystals or evidence of sausserization (Eggleton et al., 1987; Glassman, 1982).

Of the samples collected from the field, three were chosen for examination by electron microprobe based on apparent amount of weathering and the presence void spaces across the surface of the thin section. These three samples were carbon coated and examined by a JEOL JEM 1200-EX electron microprobe (EMP) at UCLA for the elemental abundances of individual plagioclase crystals. A total of 14 points were collected from plagioclase crystals from the first thin section. Three images were taken from this sample and of those 2 data points were taken from the first image, 3 from the second and 9 from the third. Nine data points total were taken from one image in the second sample. Six data points were taken from the third sample. For each of these samples, points were collected from plagioclase crystals at varying distances from a void

feature. Samples were also taken at a distance from any void feature. The rock samples were crushed and powdered and ICP-OES analyses were completed on all samples collected. The major elements were determined by ICP-OES at ActLabs and trace elements by ICP-MS at Union College.

## RESULTS

The rocks examined by thin section were found be composed of volcanic glass, plagioclase, pyroxenes, olivine and opaque minerals. In hand sample, the outer portions of rock collected appeared weathered. Discoloration to a reddish brown or brown was visible, and the outer rock was more friable than the inner portion of the rock. In thin section, the rocks also appeared to be weathered to a slight degree. The interstitial glass was becoming yellowed in some higher silica samples and olivine and pyroxene phenocrysts exhibited a rough texture on the surface and edges of the crystals in some samples. Plagioclase phenocrysts also appeared to be scalloped around the edges and some appeared dusty in the central area of the crystal. The birefringence was also lower than expected for the plagioclase phenocrysts. They were not the first order grays and whites expected for plagioclase under cross polarized light, but yellowed.

The fine microcrysts that make up the groundmass of many of the rocks were foggy and dusty looking indicating that weathering had begun in that area. The alteration of the groundmass was most noticeable close to voids and decreased away from them as expected (Glassman, 1982). These indicators of weathering were observed mostly in the samples closer to the surface of the rock, but also alongside cracks and vesicles (Fig. 2). As the minerals are weathered, cracks and void space forms within the rock that allow even more water to flow through the rock and continue alteration (Glassman, 1982). Even so, some samples were found with cracks running directly through a plagioclase crystal with no apparent weathering taking place (Fig. 3).

The three samples that were examined by EMP did not exhibit a high degree of chemical weathering

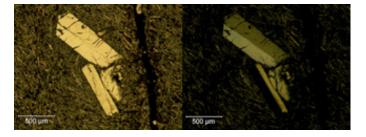


Figure 2: Photomicrographs in plane and crossed-polarized light of Plagioclase phenocrysts in andesite. Notice apparent weathering of plagioclase near fracture.

of plagioclase. It was expected that ratios of K/Na and K/Ca would increase with distance from voids. There was only a very weak relation between the removal of Na and Ca and distance from voids. Plagioclase weathering rate is expected to increase with An content of the plagioclase crystal (White and Brantley, 2003). This correlation was not detected at all. There was no correlation between An content in a plagioclase phenocryst and distance to a void found in these rocks. Backscatter electron (BSE) images also indicate little weathering of plagioclase crystals. Plagioclase crystals examined using BSE images exhibit little deterioration along edges of phenocrysts either near or some distance from a void space.

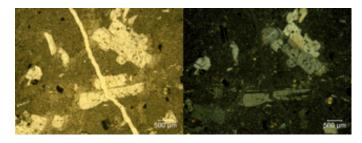


Figure 3: Photomicrographs in plane and crossed-polarized light of Plagioclase phenocrysts in dacite. Notice the fracture passes directly through phenocryst without any apparent weathering.

The ICP-OES analysis results indicated that the rocks range in composition from basalt to rhyolite. The bulk rock compositions of the samples do not indicate any significant weathering of plagioclase.

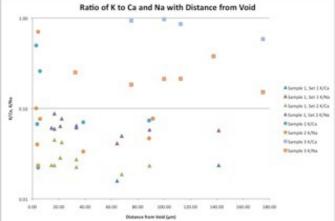


Figure 4: Graph of the ratio of K to Ca and Na with distance from a void in a basalt, andesite and rhyolite.

## DISCUSSION

The composition of streams and rivers is dependent on the decomposition of rock in the area. Throughout a watershed, especially one as large as the Deschutes Basin, many, many streams and rivers are collecting ions from the rocks they pass over and through. The water chemistry may indicate a relationship with elements associated with volcanic rock (Na, Ca, K, Al, Ti, Mg, Fe), but it is not likely that the majority of these elemental traces are sourced from the plagioclase in the samples collected in this study. While the rock samples exhibit evidence for the initial stages of alteration as described above, the elemental evidence gathered through electron microprobe analysis does not support plagioclase from these rocks as the dominant source of ions in stream water.

The rocks appear to be weathered in general in hand sample, but plagioclase is not one of the first minerals to weather from a rock, it has not undergone a great degree of weathering in these rocks. The rocks sampled in this study have just begun to enter the initial stages of weathering but have not yet begun to significantly weather the plagioclase. The dry climate of the field area, combined with the silicic composition of some of the rocks and the relatively young age of all of the rocks has not allowed for any significant weathering of the plagioclase.

	Central Oregon								
Sample Data Point	1-1 1	1-1 2	1-2 3	1-2 4	1-2 5	1-2 6	1-2 7	1-2 8	
TiO2	0.080	0.060	0.094	0.146	0.136	0.069	0.123	0.109	
SiO2	53.060	53.610	52.864	55.365	54.046	52.945	53.812	53.305	
AI2O3	29.320	28.800	29.274	27.216	28.604	29.037	28.406	28.844	
FeO	0.620	0.860	0.770	1.094	0.691	1.027	0.936	0.864	
MnO	0.020	0.010	0.039	0	0.051	0.008	0.550	0.017	
MgO	0.200	0.170	0.160	0.188	0.140	0.166	0.133	0.198	
CaO	12.020	11.630	12.177	10.232	11.211	12.027	11.382	11.814	
Na2O	4.450	4.600	4.411	5.388	4.824	4.477	4.875	4.564	
K20	0.160	0.240	0.196	0.372	0.274	0.244	0.268	0.237	
SO3	0.080	0.010	0.015	0	0.024	0	0.064	0.047	
TOTAL	100	100	100	100	100	100	100	100	
Sample	1-2	1-2	1-2	1-2	1-2	2	2	2	
Data Point	9	10	11	1-2	13	14	15	16	
TiO2	0.074	0.160	0.341	0.094	0.146	0.408	0.037	0.119	
SiO2	52.541	53.140	54.942	52.864	55.365	67.671	63.141	65.563	
AI2O3	52.541 29.493	28.870	26.896	52.864 29.274	27.216	19.758	23.063	65.563 19.768	
A1203 FeO	0.804	0.938	1.523	0.770	1.094	0.771	0.483	0.480	
re0 Mn0	0.804	0.938	0.036	0.039	0	0.027	0.485	0.480	
MgO	0.013	0.149	0.304	0.039	0.188	0.027	0.001	0.008	
CaO	12.492	11.957	10.416	12.177	10.232	1.752	2.965	0.485	
Na2O	4.191	4.544	5.123	4.411	5.388	8.364	9.628	8.311	
K20	0.242	0.241	0.402	0.196	0.372	0.748	0.659	5.220	
\$03	0	0	0.018	0.015	0	0.061	0.024	0.045	
TOTAL	100	100	100	100	100	100	100	100	
								3	
Sample	2	2	2	2	2	2			
Sample Data Point	2 17	2 18	2 19	2 20	2 21	2 22	3 23	24	
Data Point									
Data Point TiO2 SiO2	17	18	19	20	21	22	23	24	
Data Point TiO2 SiO2	<b>17</b> 0.086	<b>18</b> 0.085	<b>19</b> 0.079	<b>20</b> 0.126	<b>21</b> 0.127	<b>22</b> 0.130	<b>23</b> 0.041	<b>24</b> 0.018	
Data Point TiO2 SiO2 Al2O3 FeO	17 0.086 55.601 27.989 0.703	18 0.085 59.829 24.987 0.847	<b>19</b> 0.079 66.030	20 0.126 60.233 24.665 0.698	<b>21</b> 0.127 60.314 24.355 0.784	22 0.130 58.255 25.923 0.793	<b>23</b> 0.041 65.097	24 0.018 65.200	
Data Point TiO2 SiO2 Al2O3 FeO	<b>17</b> 0.086 55.601 27.989	18 0.085 59.829 24.987	19 0.079 66.030 19.027	<b>20</b> 0.126 60.233 24.665	<b>21</b> 0.127 60.314 24.355	<b>22</b> 0.130 58.255 25.923	<b>23</b> 0.041 65.097 21.782	24 0.018 65.200 21.613	
Data Point TiO2 SiO2 Al2O3 FeO MnO	17 0.086 55.601 27.989 0.703 0.045 0	18 0.085 59.829 24.987 0.847	19 0.079 66.030 19.027 0.507	20 0.126 60.233 24.665 0.698	<b>21</b> 0.127 60.314 24.355 0.784	22 0.130 58.255 25.923 0.793	23 0.041 65.097 21.782 0.237	24 0.018 65.200 21.613 0.307 0 0	
Data Point TiO2 SiO2 Al2O3 FeO MnO MgO CaO	17 0.086 55.601 27.989 0.703 0.045 0 8.065	18 0.085 59.829 24.987 0.847 0 0.025 5.280	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145	20 0.126 60.233 24.665 0.698 0 0.004 4.612	<b>21</b> 0.127 60.314 24.355 0.784 0	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439	24 0.018 65.200 21.613 0.307 0 0 2.053	
Data Point TiO2 SiO2 Al2O3 FeO MnO MgO CaO Na2O	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035	
Data Point TiO2 SiO2 Al2O3 FeO MnO MgO CaO Na2O K2O	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357 0.153	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993 8.139	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708	
Data Point TiO2 SiO2 Al2O3 FeO MnO CaO CaO Na2O K2O SO3	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357 0.153 0	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993 8.139 0.044	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point TTO2 SIO2 Al2O3 FeO MnO MgO CaO Na2O K2O SO3 TOTAL	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357 0.153 0 100	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993 8.139 0.044 100	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708	
Data Point           TiO2           SiO2           Al203           FeO           MnO           MgO           CaO           Na2O           K2O           SGO3           TOTAL           Sample	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357 0.153 0 100 3	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 3	19 0.079 66.030 19.027 0.031 0.004 0.145 5.993 8.139 0.044 100 3	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 3	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
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Data Point TTO2 TTO2 TTO2 TTO2 TTO2 TTO2 TTO2 TTO	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357 0.153 0 100 3 25 0.033 68.429	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 3 26 0.056 65.643	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993 8.139 0.044 100 3 27 0 65.969	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 3 28 0.077 70.249	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point           T/O2           T/O2           Sample           Data Point           T/O2           SiO2           Al2O3	17 0.086 55.601 27.989 0.703 0.045 0 8.065 7.357 0.153 0 0.053 100 3 25 0.033 8.429 19.805	18 0.085 59.897 0.847 0 0.025 5.280 8.552 0.305 0.090 100 3 26 0.056 65.643 21.653	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993 8.139 0.044 100 3 27 0 65.969 21.152	20 0.126 60.233 24.665 0.698 0 0 0.004 4.612 9.363 0.279 0.021 100 3 28 0.077 70.249 18.467	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point TTO2 TTO2 TTO2 TTO2 TTO2 TA Data Point TTO2 StO2 StO3 TTO2 Sample Data Point TTO2 StO2 StO3 TTO2 StO3 TTO2 StO3 Sto3	17 0.086 55.601 27.989 0.703 0.703 0.703 0.703 0.703 0.703 0.7357 0.153 0 100 100 3 3 5 0.033 68.429 19.805 0.268	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 100 3 3 6 0.056 65.643 21.653 0.525	19 0.079 66.030 19.027 0.507 0.507 0.507 0.507 0.507 0.507 0.507 0.507 0.507 0.415 100 115 0 0.044 100 0 3 27 0 65.590 0 21.152 0.310	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 3 3 8 0.077 70.249 18.467 0.436	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point TTO2 TTO2 STO2 AI2O3 FeO MnO MgO CGO CGO CGO SO3 TOTAL Sample Data Point TTO2 STO2 STO2 AI2O3 FeO MnO	17 0.086 55.601 27.989 0.703 0.703 0.703 0.703 0.703 0.703 0.703 0.703 0.703 0.703 0.703 0.703 0.705 0.703 0.705 0.703 0.705 0.7	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 3 0.055 65.643 21.653 0.525 0.214	19 0.079 66.030 19.027 0.507 0.507 0.507 0.001 0.004 0.145 5.993 8.139 0.044 100 3 27 0 65.969 21.152 0.310 0 0	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 3 28 28 0.077 7.0.249 18.467 0.436 0.013	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point TTO2 TTO2 TTO2 TTO2 SiO2 Al203 Mn0 Mg0 Ca0 Mg0 Ca0 Mg0 Ca0 Mg0 Sample Data Point TTO2 Sample Data Point TTO2 Sample Data Point TTO2 SiO2 Al203 SiO2 Al203 Sample Data Point TTO2 SiO2 Al203 SiO2 SiO3 SiO2 SiO3 SiO2 SiO3 SiO3 SiO2 SiO3 SiO2 SiO3 SiO2 SiO3 SiO2 SiO3 Si	17 0.086 55.601 27.989 0.703 0.045 0.703 0.153 0 100 100 100 100 100 100 100 100 100	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 100 3 26 65.643 21.653 0.525 0.525 0.224	19 0.079 66.030 19.027 0.507 0.031 0.004 0.145 5.993 8.139 0.044 100 100 <b>3</b> 27 0 65.569 21.152 0.310 0 0	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 100 100 3 28 0.077 70.249 18.467 0.436 0.013	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point TTO2 TTO2 SIO2 Al2O3 FeO MnO MgO CaO No2O CaO No2O KaZO SO3 TOTAL TTO2 Sample Data Point TTO2 SiO2 Al2O3 FeO MnO MgO CaO	17 0.086 55.601 27.989 0.703 0.45 0 8.065 7.357 0.153 0 100 100 3 3 5 0.033 68.429 19.805 0.268 0.010 0.208	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 100 3 3 6 5.63 0.056 65.633 0.525 0.214 0.024 0.224	19 0.079 66.030 19.027 0.507 0.507 0.004 0.145 5.993 8.139 0.044 100 3 27 0 65.969 21.152 0.310 0 0 0 1.864	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 3 8 0.077 7.0249 18.467 0.436 0.436 0.436 0.013 0.003 1.158	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point TTO2 TTO2 SIO2 AI2O3 FeO MnO MgO CaO Na2O SO3 TOTAL Data Point TTO2 SIO2 SIO2 SIO2 SIO2 SIO2 SIO2 SIO2 SI	17 0.086 55.601 27.989 0.703 0.703 0.45 0 8.065 7.357 0.153 0 100 3 2 0.033 68.429 19.805 0.268 0.010 0.268 0.010 0.268 0.010 0.265 0.265 0.101 0.265	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 100 3 6 5.643 21.653 0.525 0.214 0.024 2.117 8.206	19 0.079 66.030 19.027 0.507 0.507 0.004 0.145 5.993 8.139 0.044 100 3 7 7 0 65.969 21.152 0.310 0 0 0 1.864 9.163	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 0.021 100 3 8 0.077 70.249 18.467 0.436 0.013 0.003 1.158	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point           TTO2           TTO2           Store           Mn0           Mg0           Ca0           Na2O           K20           S03           TOTAL           SiO2           Al203           SiO2           Sample           Data Point           TTO2           SiO2           Al203           Fe0           Mn0           Mg0           Ca0           Na2O           K2O	17 0.086 55.601 27.989 0.703 0.045 7.357 0.153 0 100 100 100 100 100 100 100 100 100	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 100 3 26 65.643 21.653 0.525 0.214 0.024 2.117 8.206	19 0.079 66.030 19.027 0.031 0.004 0.145 5.993 8.139 0.044 100 100 3 27 0 6 5.5969 21.152 0.310 0 0 1.864 9.163 1.499	20 0.126 60.233 24.665 0 0.004 4.612 9.363 0.279 0.021 100 100 100 3 28 0.077 70.249 18.467 0.436 0.003 1.158 7.169 2.416	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	
Data Point           T/O2           SiO2           Al203           FeO           MnO           MgO           CaO           Na2O           K2O           SO3           TOTAL           Sample	17 0.086 55.601 27.989 0.703 0.45 0 8.065 7.357 0.153 0 100 100 3 3 5 0.033 68.429 19.805 0.268 0.268 0.268 0.265 0.268	18 0.085 59.829 24.987 0.847 0 0.025 5.280 8.552 0.305 0.090 100 100 3 6 5.643 21.653 0.525 0.214 0.024 2.117 8.206	19 0.079 66.030 19.027 0.507 0.507 0.004 0.145 5.993 8.139 0.044 100 3 7 7 0 65.969 21.152 0.310 0 0 0 1.864 9.163	20 0.126 60.233 24.665 0.698 0 0.004 4.612 9.363 0.279 0.021 100 0.021 100 3 8 0.077 70.249 18.467 0.436 0.013 0.003 1.158	21 0.127 60.314 24.355 0.784 0 0.018 5.559 8.428 0.351 0.064	22 0.130 58.255 25.923 0.793 0.031 0.035 7.283 6.984 0.482 0.085	23 0.041 65.097 21.782 0.237 0.018 0.021 2.439 9.067 1.225 0.074	24 0.018 65.200 21.613 0.307 0 0 2.053 9.035 1.708 0.067	

Electron Microprobe Analysis of Volcanic Rocks of Deschutes Basin, Central Oregon

Table 1: Table of elemental percentages in a basalt, andesite and rhyolite. Note that these values are normalized.

### **CONCLUSION AND FUTURE WORK**

Since it would follow that the source of elemental traces found in streams in the Deschutes Basin is not the plagioclase from the quaternary volcanic rock studied here, an interesting extension to this project would be to compare rocks of the dry Deschutes Basin to rocks of similar composition to from the wet climate of the western part of Oregon. Would significant weathering of the rocks of the same composition and age occur in an environment receiving more precipitation? Plagioclase is slow to begin weathering. It will not begin weathering until most of the other minerals have undergone some degree of weathering. This is the reason that there is very little EMP evidence for weathering. The rocks are too young. Would older rocks from the same area in the Deschutes Basin exhibit a greater degree of weathering than the rocks studied here? The results of this study provide an initial position for more investigation of weathering of plagioclase in young rocks of a dry climate.

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